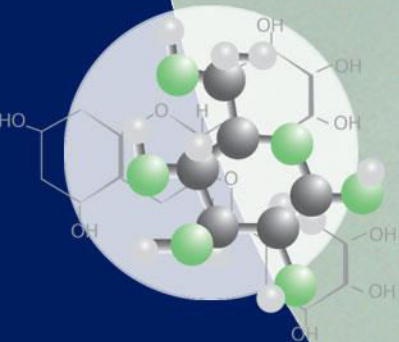


GI physiology and need for biorelevant media

Disso India 2017

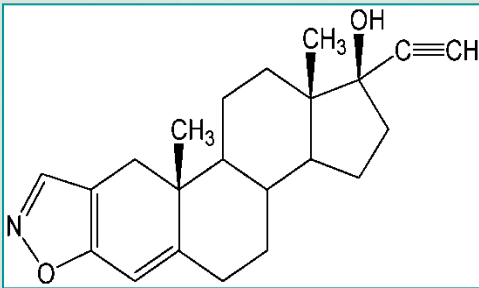
Workshop
Mumbai

Prof. Dr. Jennifer Dressman

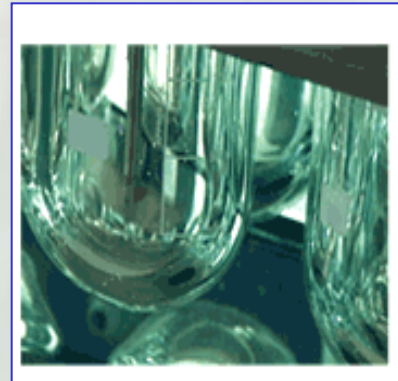


GI physiology and dissolution testing

- **What factors influence release from drug products?**
 - **The properties of the drug**
 - **The quality and design of the drug product**
 - **The conditions under which the test is run**



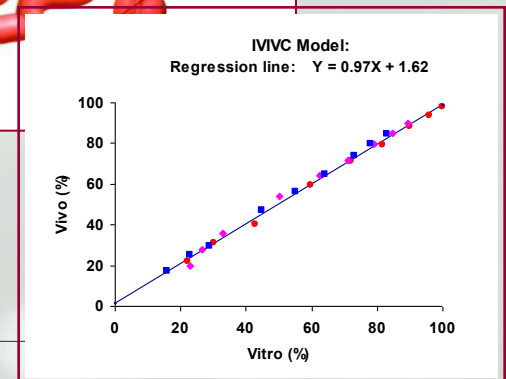
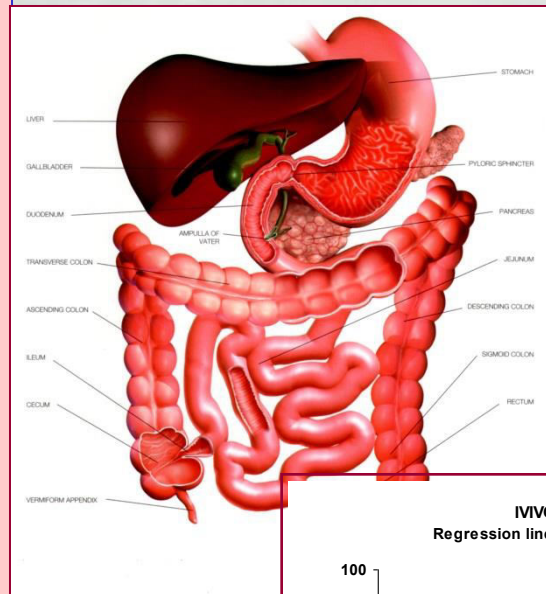
Picture of
Mortar and
pestle



GI physiology and dissolution testing

Hypothesis:

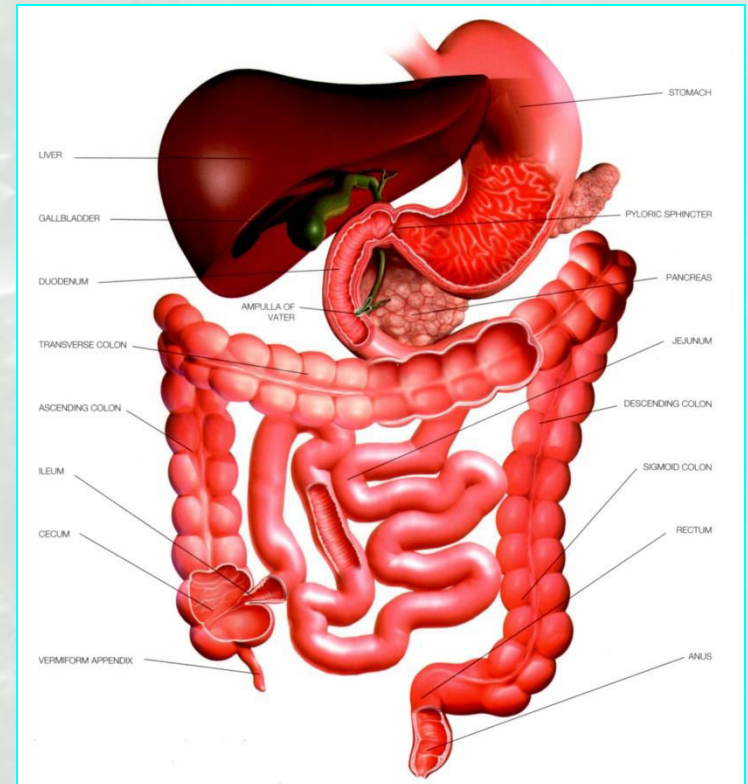
the closer the dissolution test conditions to the physiology, the better the chances of predicting *in vivo* performance



GI physiology and dissolution testing

THREE important considerations:

- 1) **WHERE** in the GI tract is drug released from the dosage form
- 2) **HOW LONG** does the dosage form have to release the drug
- 3) **COMPOSITION** of the fluids into which drug is released



GI physiology and dissolution testing

- 1) **WHERE** in the GI tract is drug released from the dosage form? This will vary with the **drug product** e.g.
 - 1) Immediate release dosage forms
 - 2) Enteric coated dosage forms
 - 3) Extended release dosage forms
 - 4) Pulsatile delivery....

The site(s) of release and/or % released at each site of release are often also dependent on whether the dosage form is given before or after a meal, so the dissolution test should reflect the **dosing conditions**

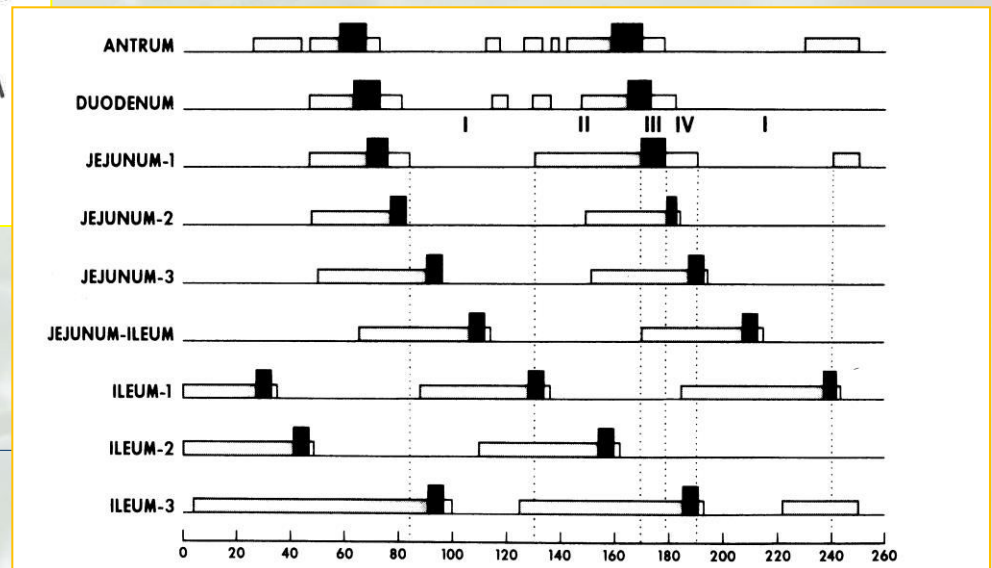
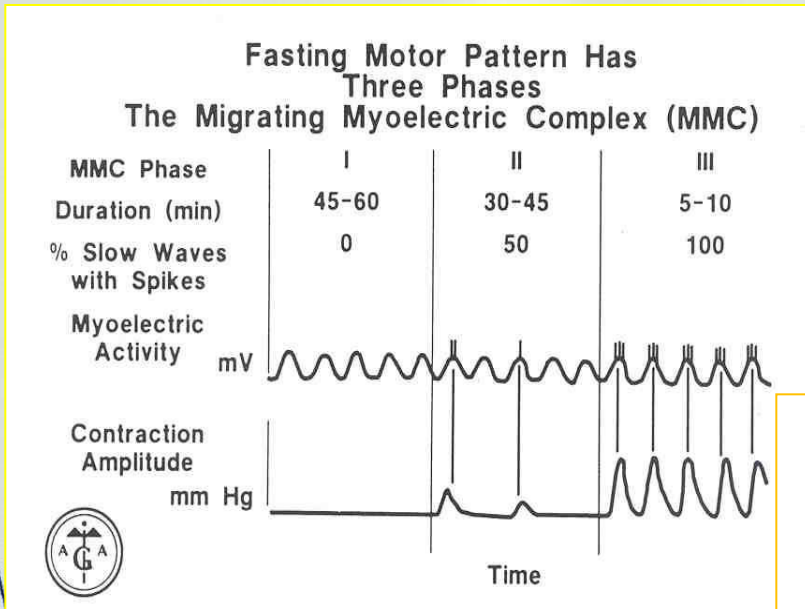
GI physiology and dissolution testing

- 1) **HOW LONG** does the dosage form have to release the drug?
 - The drug must be released before or at its site(s) of absorption, otherwise release will not result in absorption. So it is important to understand the **permeability** of the drug at various points in the gut.
 - The passage of the dosage form through the stomach depends on **unit size** and **prandial state**.
-

GI physiology and dissolution testing

1) **HOW LONG** does the dosage form have to release the drug?


In the **fasted state**, motility in the upper GI tract is cyclical and passage is size-independent



GI physiology and dissolution testing

- 1) **HOW LONG** does the dosage form have to release the drug?

In the **fed state**,
passage of bigger
units may be
considerably delayed

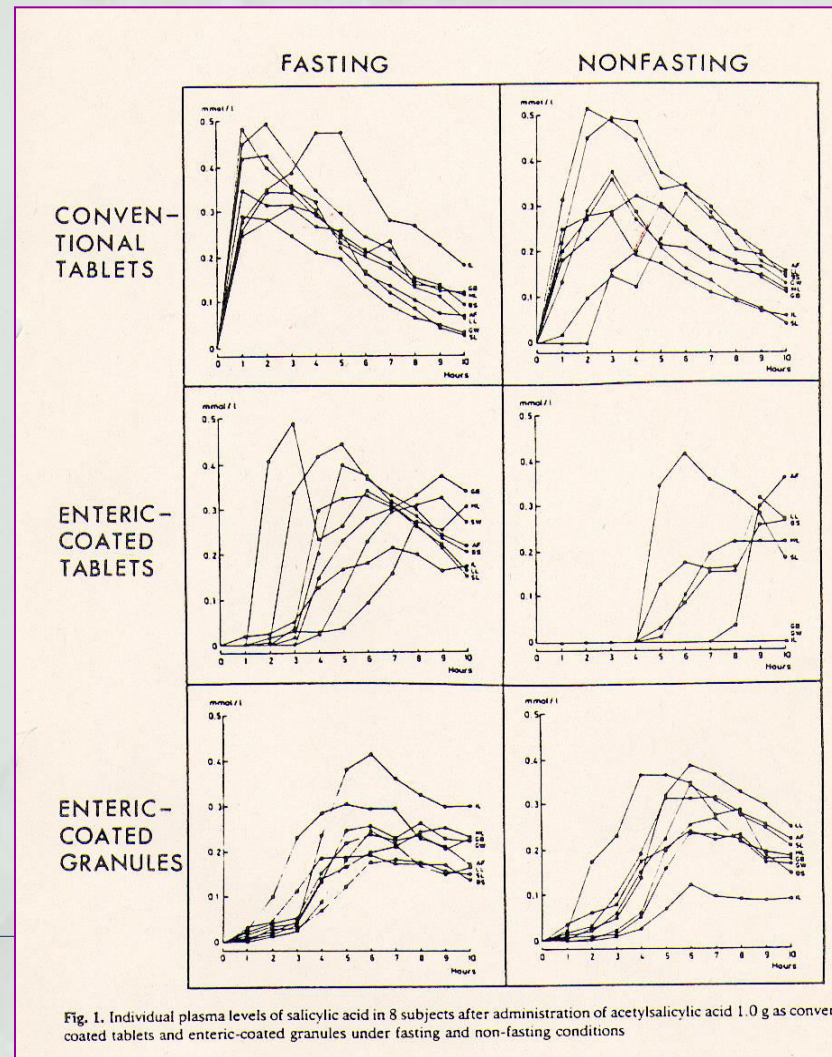


Picture showing gastric
motility patterns in the
fed state

GI physiology and dissolution testing

1) **HOW LONG** does the dosage form have to release the drug?

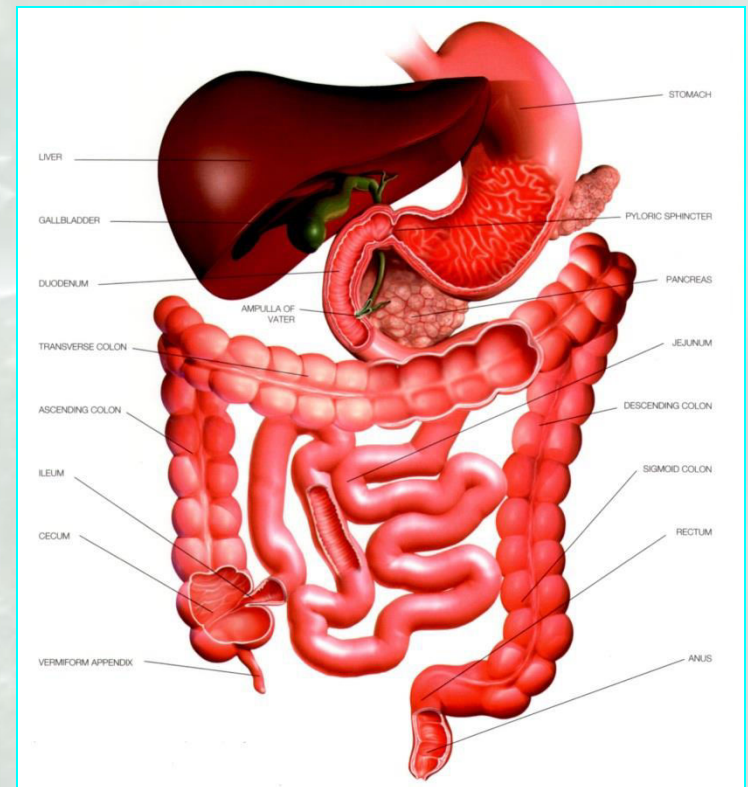
These effects can lead to huge differences in the plasma profiles



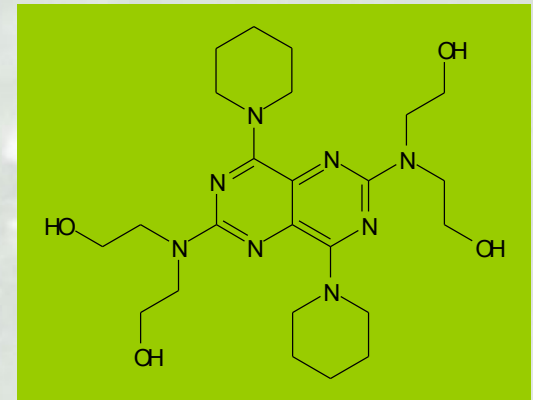
GI physiology and dissolution testing

COMPOSITION of the fluids into which drug is released

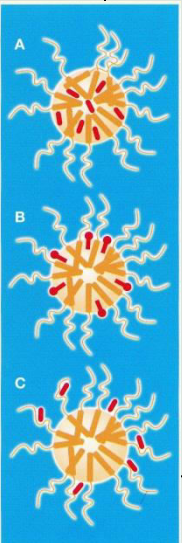
The foods and drinks we consume, gastric juices, bile, pancreatic juices, bacterial fermentation as well as water re-uptake all combine to influence the composition of the GI fluids at various points in the gut.



Solubility of Dipyridamole ($\mu\text{g/ml}$) in buffers and human aspirates

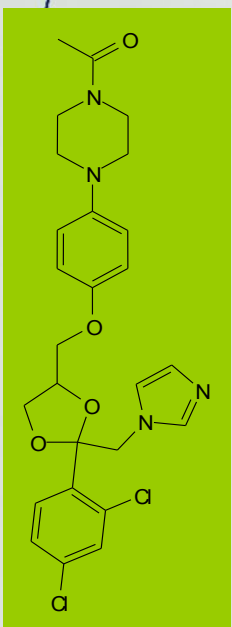


- pH 5 **60** (Kohri et al. IJP 1992)
- pH 6 **13** (Kohri et al. IJP 1992)
- pH 7 **5** (Kohri et al. IJP 1992)



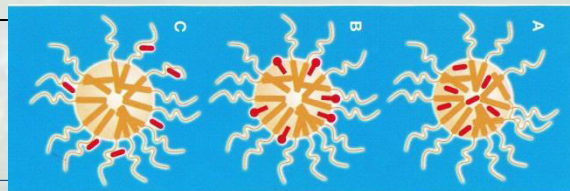
<i>HIF fasted</i> (pH 6.7)	22.5
<i>HIF fed</i> 30 and <i>HIF fed</i> 60 (pH 6.5)	160
<i>HIF fed</i> 120 (pH 5.8)	173
<i>HIF fed</i> 180 (pH 4.9)	254

Solubility of Ketoconazole ($\mu\text{g/ml}$) in buffers and human aspirates



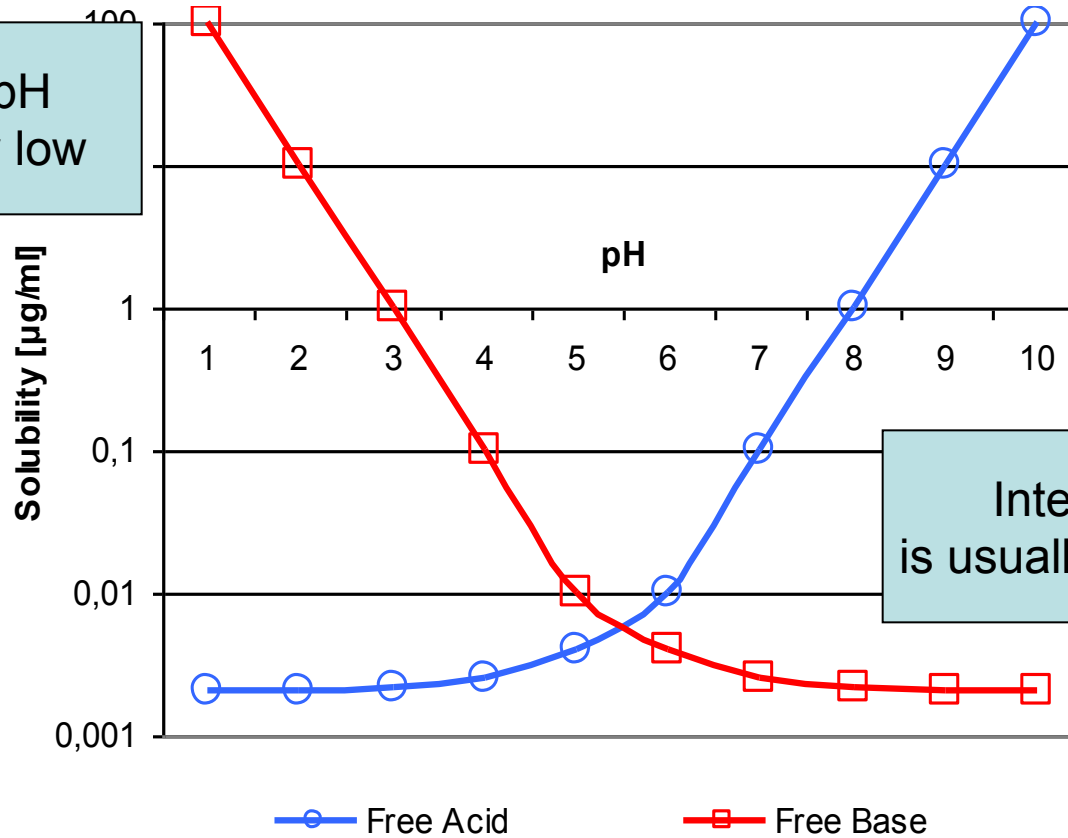
- pH 5 ~90 (Esclusa-Diaz et al. IJP 1996)
- pH 6 ~13 (Esclusa-Diaz et al. IJP 1996)
- pH 6.5 6.9 (Poelma JPP 1991)

- **HIF fasted** (pH 6.7) **28.8**
- **HIF fed** 30 and HIF fed 60 (pH 6.5) **873**
- **HIF fed** 120 (pH 5.8) **989**
- **HIF fed** 180 (pH 4.9) **476**



For weak bases and acids, solubility is highly dependent on pH

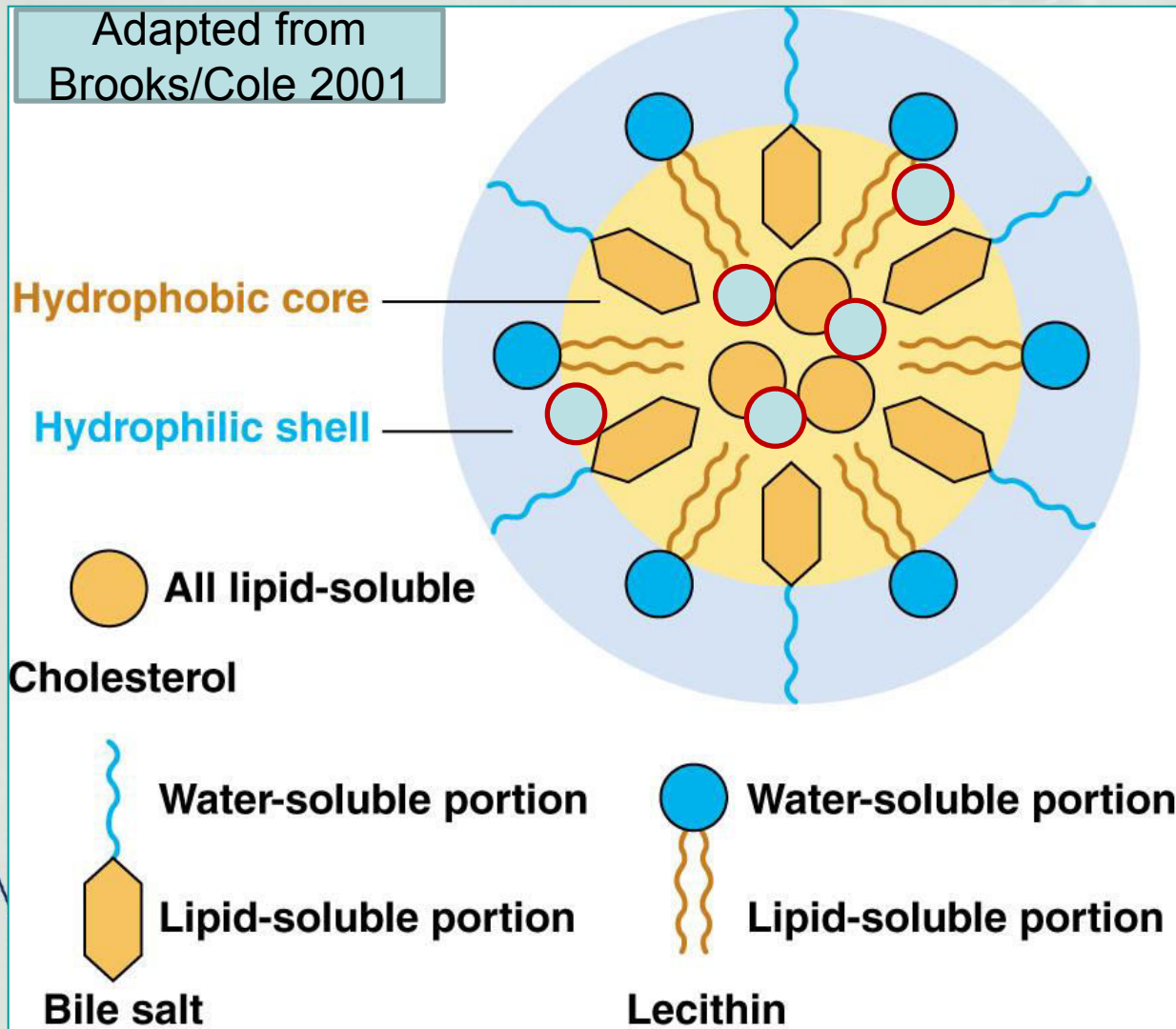
Gastric pH is usually low



Intestinal pH is usually near neutral

Solubilization by mixed micelles in the bile

Adapted from
Brooks/Cole 2001



Important for
lipophilic drugs



GI physiology and dissolution testing

COMPOSITION of the fluids into which drug is released

The foods and drinks we consume, gastric juices, bile, pancreatic juices, bacterial fermentation as well as water re-uptake all combine to influence the composition of the GI fluids at various points in the gut.

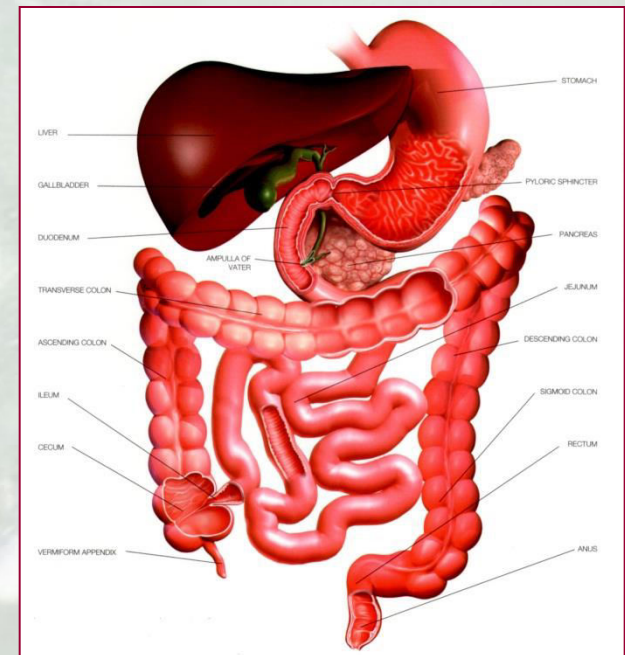
Not only the **drug**, but also the **excipients**, can have dissolution/release characteristics that are dependent on the composition.

GI-appropriate media composition and volume: „biorelevant“ dissolution media

1. Fasted state

- Stomach:
 - FaSSGF: simulates reduced surface tension in the stomach
- Small intestine:
 - FaSSIF to simulate basal bile secretion in upper SI

Vertzoni et al. EJPB 2005,
Dressman et al. Pharm.Res. 1998



In vitro simulation of the gastric contents: **preprandial (FaSSGF)**

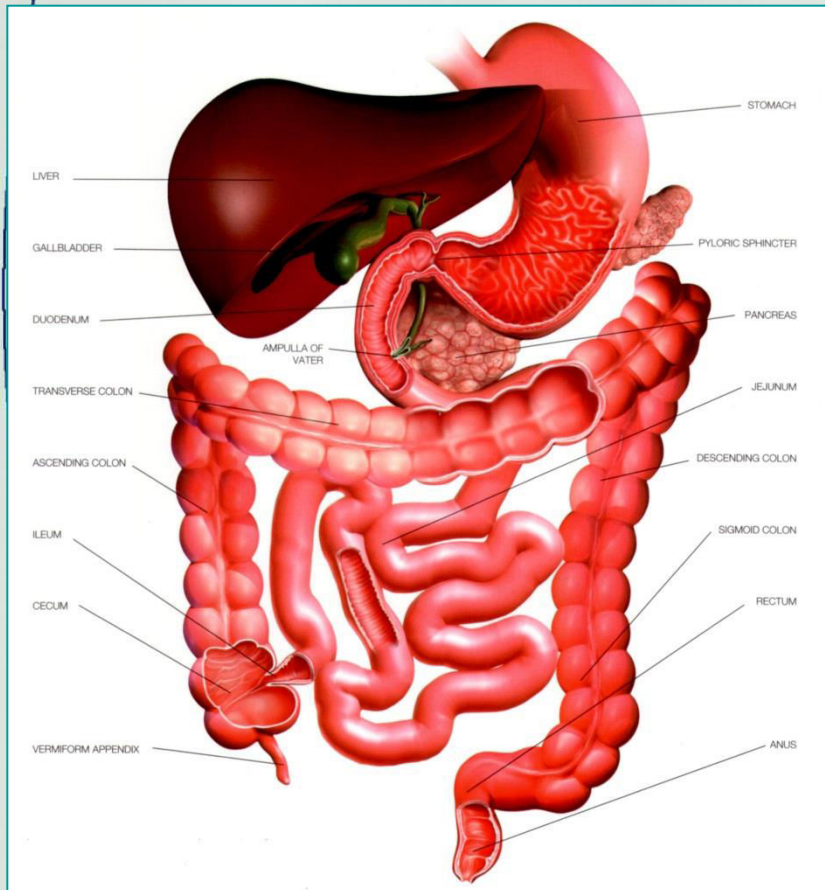
HCl	<i>q.s. pH 1.6</i>
<i>Pepsin</i>	<i>0.1 g</i>
<i>Sodium Taurocholate</i>	<i>80 μM</i>
<i>Lecithin</i>	<i>20 μM</i>
Sodium chloride	<i>34.2 mM</i>
Distilled Water	<i>q.s 1,000 ml</i>

- *Vertzoni et al. Eur J Pharm Biopharm 60 (2005) 413-417*

In vitro simulation of the small intestine contents: **preprandial** (FaSSIF-V2)

Maleic acid		19.12 mM
Sodium taurocholate		3 mM
Lecithin		0.2 mM
NaCl		68.62 mM
NaOH		34.80 mM
Distilled Water	qs	500 ml
<hr/>		
<i>pH</i>		6.5
<i>Osmolality</i>		180 \pm 10 mOsm
<i>Buffer Capacity</i>		10 \pm 2 mEq/L/pH unit
<hr/>		

Simulation of the fed state in the upper GI tract



Picture of food

GI-appropriate media composition and volume: „biorelevant“ dissolution media

2. Fed State

- **Stomach:**
 - FeSSGF: Milk/buffer pH 5 combination to simulate gastric conditions after a standard breakfast
- **Small intestine:**
 - „FeSSIF-V2“ to simulate postprandial bile secretion, lipolysis products, increased buffer capacity and osmolality in upper SI after food intake



Picture of
bile salt
micelle

**in vitro simulation of the gastric
contents: *postprandial* (FeSSGF)**

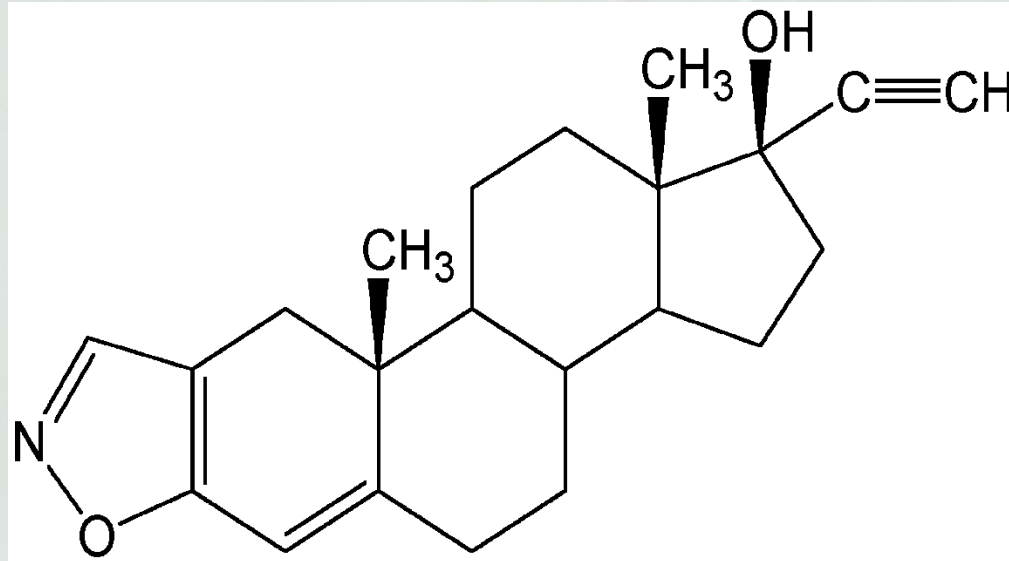
Acetic acid	17.12 mM
Sodium acetate	29.75 mM
Sodium chloride	237.02 mM
Milk: Buffer	1:1
NaOH/HCl	q.s. pH 5

This medium has a pH of 5, Osmolality 400 mOsmol/kg, buffer capacity 25 mmolE/l/ Δ pH

in vitro simulation of the small intestinal contents: **postprandial (FeSSIF-V2)**

<i>Sodium taurocholate</i>	<i>10 mM</i>
<i>Lecithin</i>	<i>2 mM</i>
<i>Glycerol monooleate</i>	<i>5 mM</i>
<i>Sodium oleate</i>	<i>0.8 mM</i>
<i>Maleic acid</i>	<i>55 mM</i>
<i>Sodium hydroxide</i>	<i>81.65 mM</i>
<i>NaCl</i>	<i>125.5 mM</i>
<i>Distilled Water</i>	<i>qs 1 Liter</i>
<hr/>	
<i>pH</i>	<i>5.8</i>
<i>Osmolality</i>	<i>390 ± 10 mOsm</i>
<i>Buffer Capacity</i>	<i>25 mEq/L/pH unit</i>

Application of media to predicting food effects: Danazol



Aqueous solubility: 1 µg/ml

Dose: 200 mg

pKa: neutral

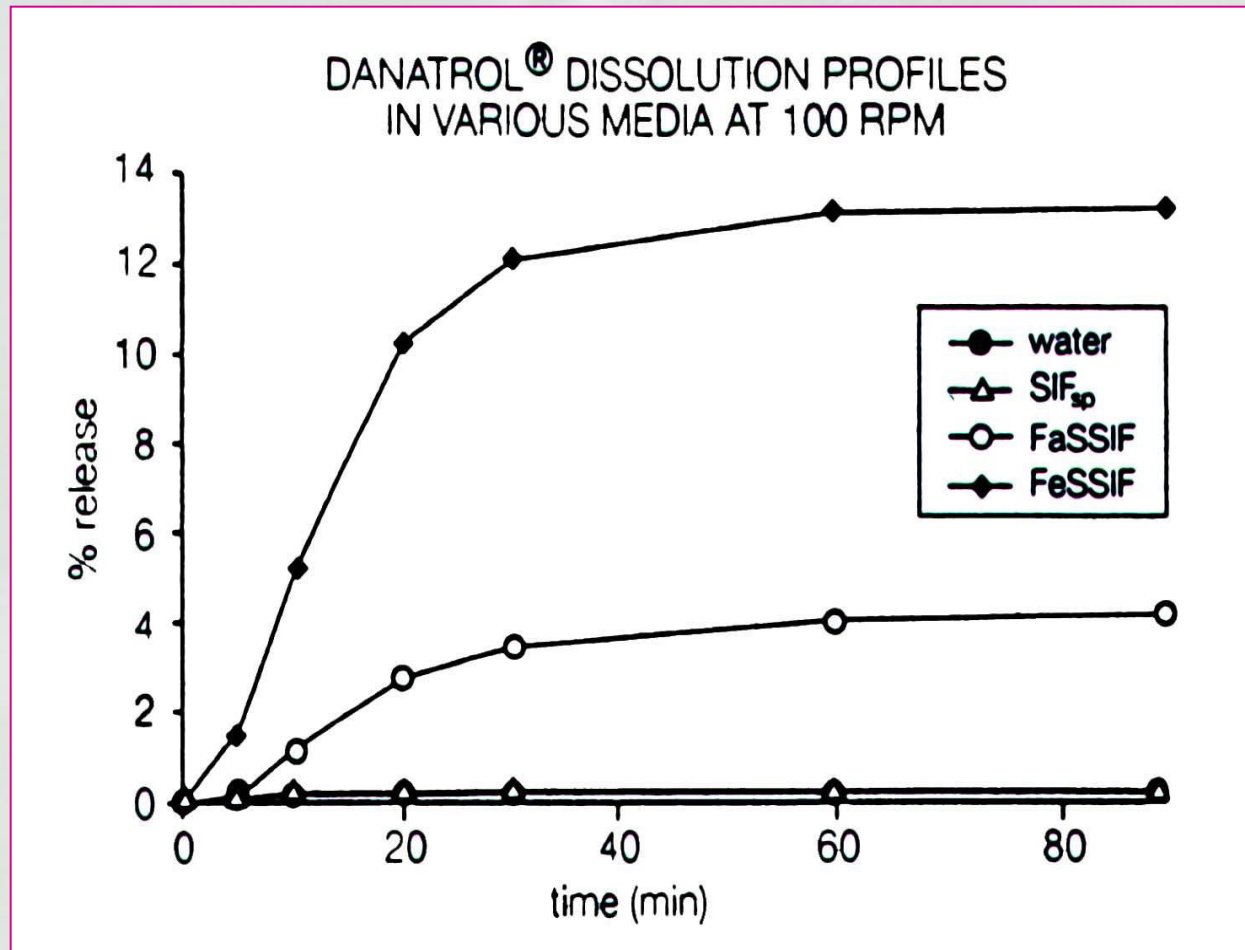
log P: 4.53

D:S 200 liters **H₂O**

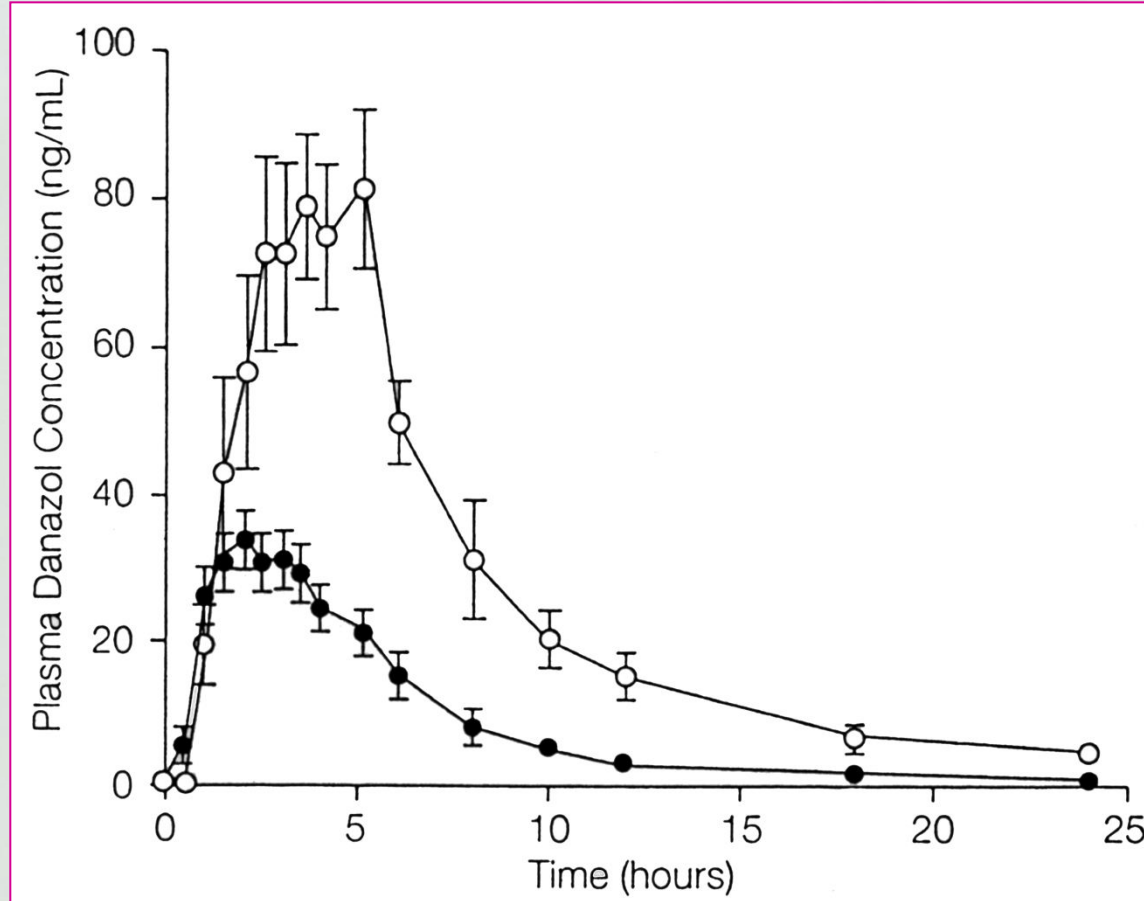
20 liters **FaSSIF**

6 liters **FeSSIF**

Danatrol dissolution profiles in various media at 100 rpm



Danazol's food effect reflects its dissolution characteristics

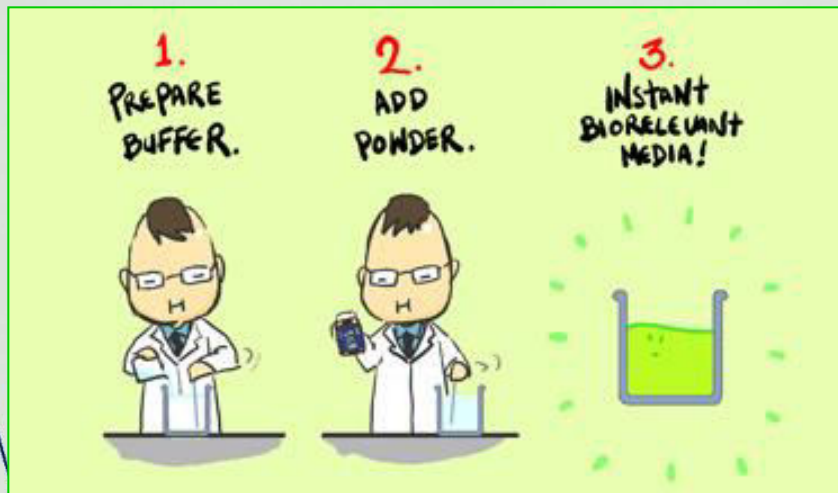


Plasma profiles of danazol after administration in the fasted (●) and fed (○) state
(from Charman *et al.*)

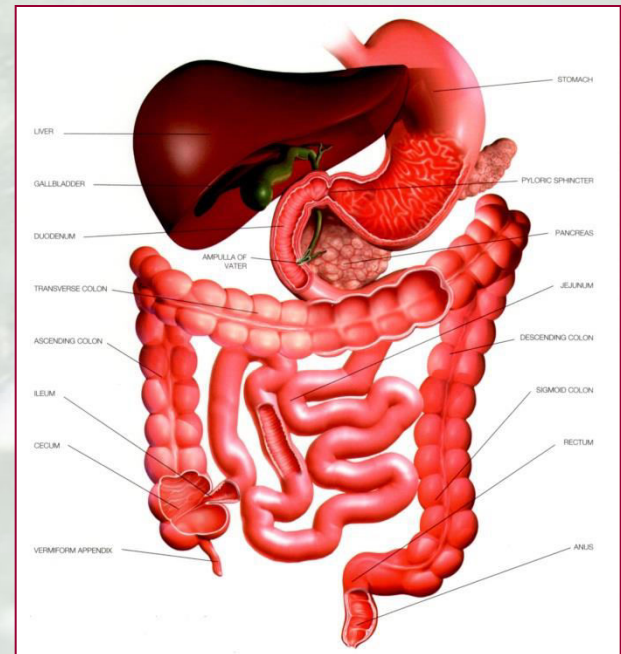
GI-appropriate media composition and volume: „biorelevant“ dissolution media

Making life easier:

Using „instant“ powders to make the biorelevant media

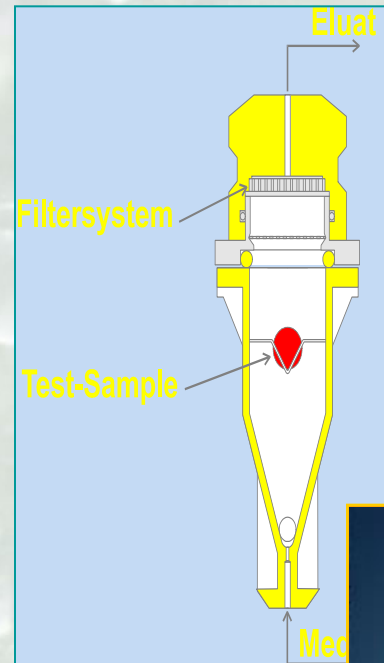


source: Biorelevant.com



Designing an appropriate Dissolution Test

- **Classify the drug substance according to BCS**
- **Choose appropriate media composition and volume**
- **Choose an appropriate apparatus**
- **Consider the hydrodynamics**
- **Determine whether de-aeration of the medium is necessary**
- **Choose an appropriate test duration**



Summary

To come up with the „right“ dissolution test for generating IVIVC, one needs to consider

- the drug's properties (solubility, permeability etc.)

- the mechanism of release of the dosage form

- dosage form dimensions

- the excipient properties

- dosing conditions in the *in vivo* study

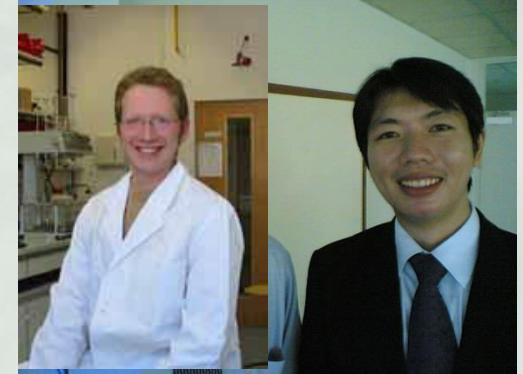
With this information, it should be possible to generate an *in vitro* profile that closely reflects the *in vivo* release profile



Acknowledgements

Niels Janssen

University of Frankfurt



Ekarat Jantratid (1975-2010)

Post-doc University of Frankfurt

Prof. Christos Reppas

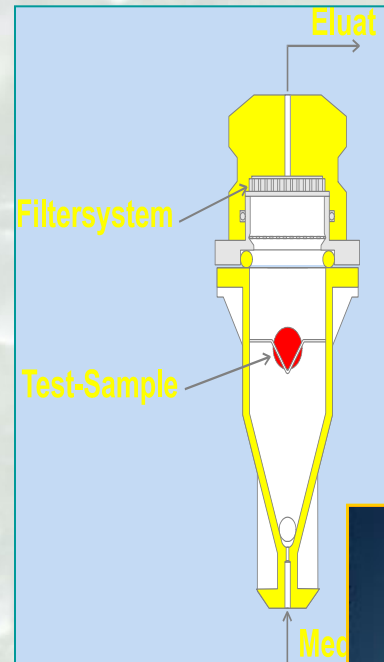
& his research group

University of Athens, Greece



Designing an appropriate Dissolution Test

- Classify the drug substance according to BCS
- Choose appropriate media composition and volume
- Choose an appropriate apparatus
- Consider the hydrodynamics
- Determine whether de-aeration of the medium is necessary
- Choose an appropriate test duration



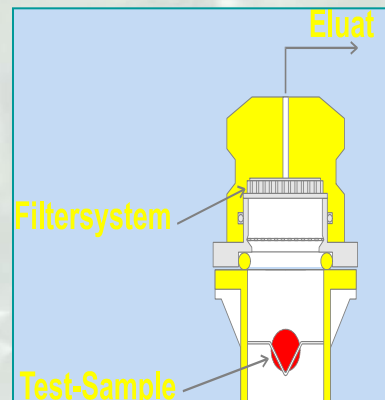
Designing an appropriate Dissolution Test

■ Notes on Media composition and volume

- 1) for *highly soluble* drugs in IR dosage forms, media composition should be simple e.g. aqueous buffer
 - 2) for *less soluble* drugs, consider biorelevant media
 - 3) if the drug is poorly soluble but highly permeable, sink conditions may be generated in the GI tract and could be considered for dissolution
 - 4) if the drug is poorly soluble and has low/moderate permeability, use of sink conditions for dissolution will likely lead to overprediction of absorption.
 - 5) Some dosage forms are far more prone to composition effects than others e.g. *enteric coated dosage form* compared to *osmotic pump*.
-

Designing an appropriate Dissolution Test

- Classify the drug substance according to BCS
- Choose appropriate media composition and volume
- Choose an appropriate apparatus
- Consider the hydrodynamics
- Determine whether de-aeration of the medium is necessary
- Choose an appropriate test duration



Dissolution apparatus

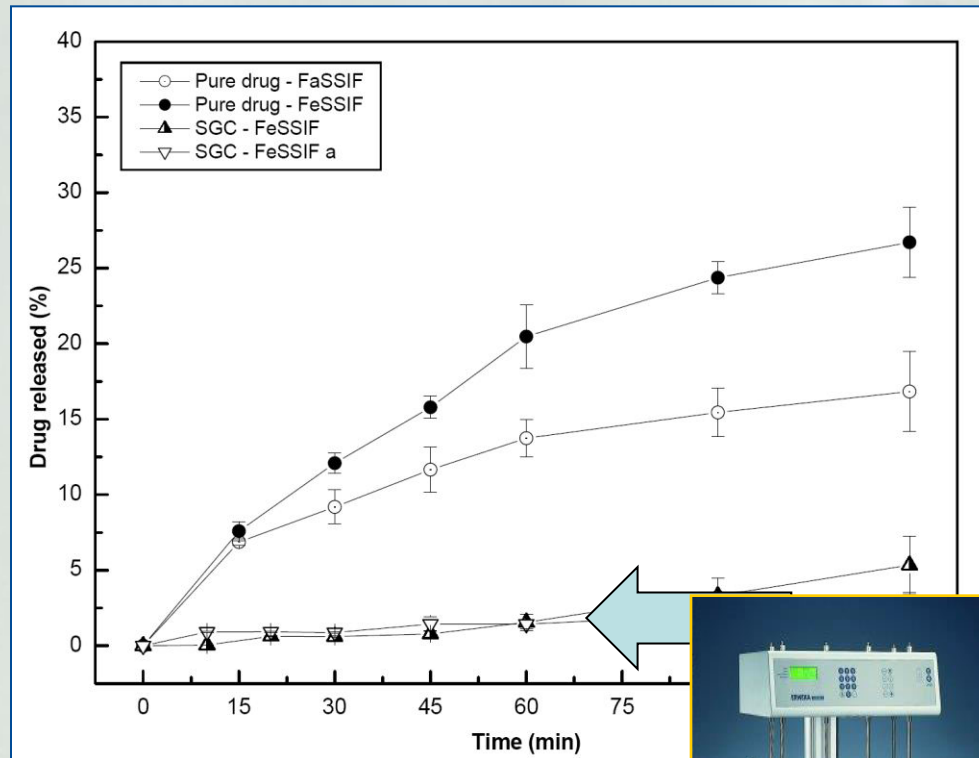
- USP* Apparatus I/II
- one vessel/unit
- basket/paddle
- volume: 500-1000 ml



Useful when one or two media will be employed

- **Less suitable for IVIVC with MR dosage forms, since IVIVC may not be possible if release testing is performed in a single medium**
- **Also unsuitable for lipid dosage forms due to poor dispersion of the lipid**

Application of the fed state media to lipid-based formulations; paddle



Dissolution in the paddle method resulted in very poor release from the formulation due to inadequate dispersion.



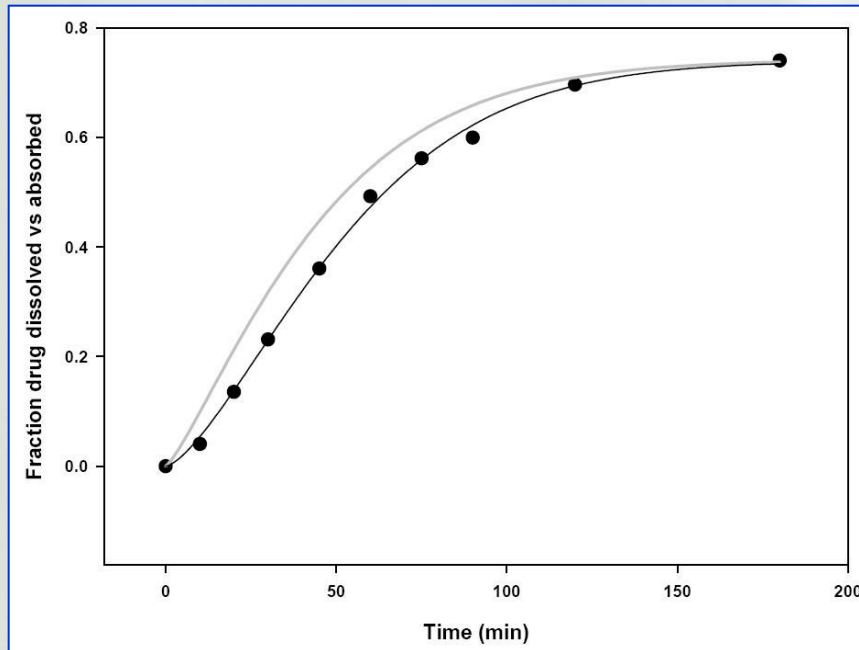
Dissolution apparatus

■ USP Apparatus III (BioDis)



- series of cylinders with sieves at each end
- volume per cylinder: 200-250 ml
- + Enables simulation of passage through the GI tract **in one test**
- + adjustment of dip-rate combined with sieve size can achieve emulsification of lipid dosage forms

Application of the biorelevant media to lipid-based formulations in the BioDis



In the BioDis, the formulation dissolved best in FeSSGF and the profile in this medium matched the absorption profile well



Application of biorelevant media in the BioDis to **MR dosage form** performance

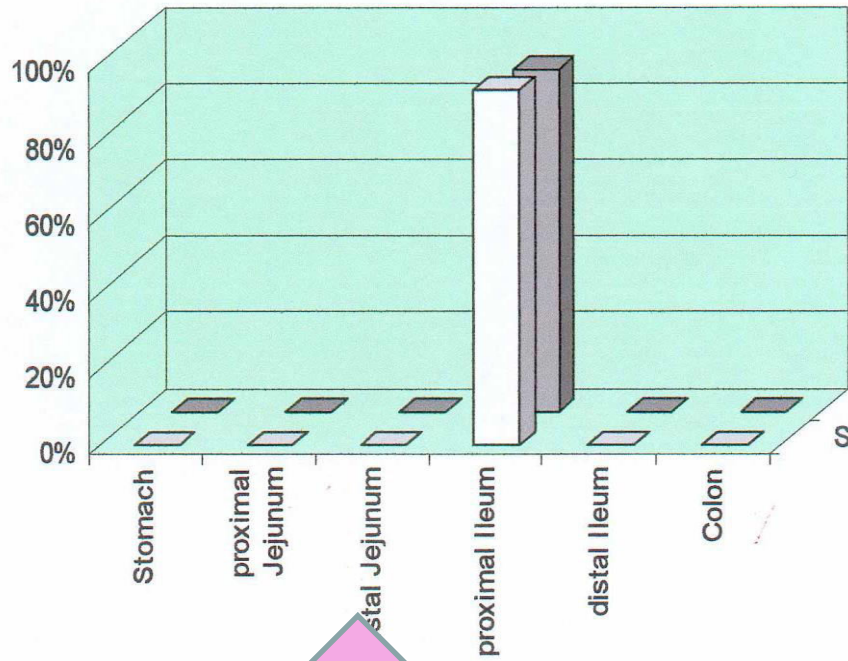
Segment of the GI tract	pH-gradient preprandial				Residence time (min)	
	blank medium	pH	biorelevant medium	pH	Tablets	Pellets
Stomach	Blank FaSSGF	1.6	FaSSGF	1.6	60	60
Duodenum/ Jejunum	Blank FaSSIF-V2	6.5	FaSSIF-V2	6.5	45	45
Jejunum/ Ileum	Blank Half-FaSSIF	7.0	Half-FaSSIF	7.0	45	45
Distal Ileum	FaSSIF-sans	7.5	FaSSIF-sans	7.5	120	120
Colon	SCoF	5.8	SCoF	5.8	480	480

Segment of the GI tract	pH-gradient postprandial				Residence time (min)	
	blank medium	pH	biorelevant medium	pH	Tablets	Pellets
Stomach	Blank FeSSGF	5.0	FeSSGF	5.0	240	120
Duodenum/ Jejunum	Blank FeSSIF-V2	5.8	FeSSIF-V2	5.8	30	45
Jejunum/ Ileum	Blank Half-FeSSIF	6.5	Half-FeSSIF	6.5	60	45
Distal Ileum	FaSSIF-sans	7.5	FaSSIF-sans	7.5	120	120
Colon	SCoF	5.8	SCoF	5.8	480	480

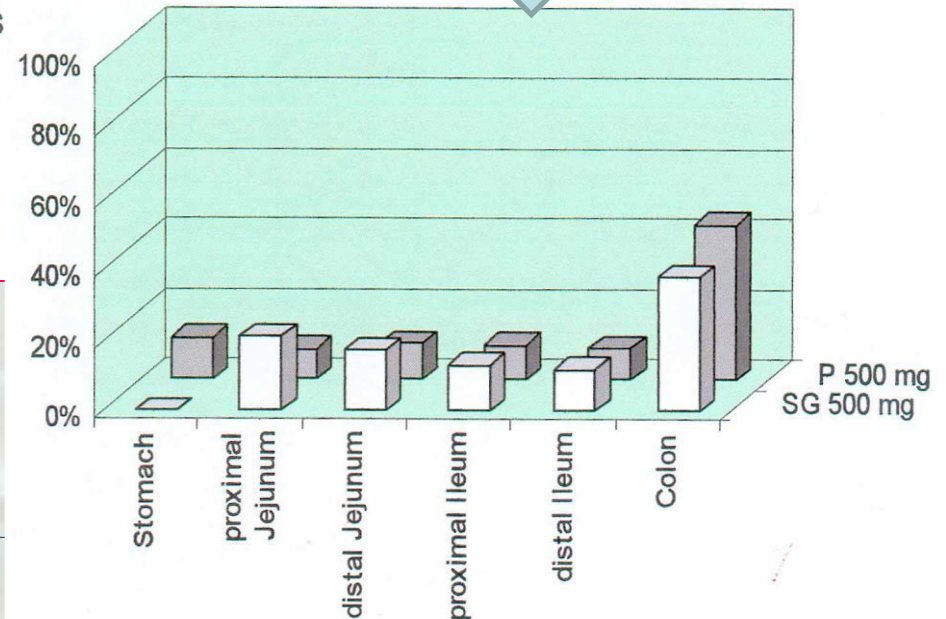
Case example: Mesalamine products

- These products are used for the therapy of Crohn's disease and ulcerative colitis in Europe
 - *Claversal®; Salofalk®*
 - Eudragit L coating (dissolves at pH > 6,0)
 - *Pentasa®*
 - Microgranulate with an Ethylcellulose coating
 - Release is diffusion driven
 - *Granustix®*
 - Eudragit L coating (dissolves at pH > 6,0) AND diffusion driven release
-

Case example: Mesalamine products



Pentasa and Granustix: Release sites in GI tract based on BioDis results



Salofalk and Claversal: Release sites in GI tract based on BioDis results